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Unit 5 Study Guide Answer Key - OCHS Chemistry

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Unit 5 Quest Answer Key - Weebly

in excess. From here figure out the grams of Al3 and you have your answer. Part C. Solution: since we have moles, we calculate directly and then convert to grams. Al and I2 stand in a two-to-three molar relationship, so 0.009456 mol of I2 uses 0.006304 mol of Al. Convert this aluminum amount to grams and subtract it from 1.20 g and that's the ...

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a) 9.5 x 10⁻¹⁰ mol/L b) 1.3 x 10⁻¹⁰ mol/L c) 6.5 x 10⁻¹¹ mol/L d) 1.9 x 10⁻⁹ mol/L e) none of these View Answer What is the molar mass of a hydrocarbon if 5.19 x 10¹⁶ molecules of the hydrocarbon ...

Chemistry Questions and Answers | Study.com

Having covered Nuclear Chemistry, Metallic Bonding and some Analysis in Unit 1 and Plastics in Unit 2, I'll be covering just about the rest of Unit 3 during Topic 7 - Metal Chemistry. Present S4 had no time in S3, but present S3 will soon be starting a Topic on Reactivity so S4 Notes refer to 'previous' work.

Chemistry Teaching Resources - National 5 N5 Chemistry

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Unit 5 - A-Level Chemistry

Unit 3 Topic 5 Isomerism & Carbonyl Chemistry ANSWERS Exercise 5 Naming Organic Compounds 1. 2-hydroxypropanoyl chloride 2. methyl propanamide 3. propanoic anhydride 4. methylpropyl ethanoate 5. N-ethylpropanamide 6. N-ethylaminoethane or diethylamine 7. 2-hydroxymethylpropanenitrile 8. 2-aminobutanoic acid 9. 10. C H H C H C Cl O H H H H C O ...

Unit 3 topic 5 ISOMERISM & CARBONYL CHEMISTRY ANSWERS

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