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change in entropy of
the ice / water system
as it came to thermal
equilibrium Suppose
0.1 kg ice at 0°C
(273K) is in 0.5kg
water at 20°C (293K).

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$$\begin{aligned}\Delta S_{\text{total}} &= \Delta S_{\text{melt}} + \Delta S_{\text{icewater}} + \Delta S_{\text{watermelt}} + \Delta S_{\text{watercooled}} \\ &= (+122.0 + 5.2 + -116.8 + -5.3) \text{ J/K} \\ &= +5.1 \text{ J/K}\end{aligned}$$

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Therefore, velocity of bullet is $(180 + 12.5) = 192.5$ m/s. To calculate velocity with which bullet with hit the thief we use concept of relative velocity. A car is

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between the formula, Eqn. 36 in the Metz-Pan paper and the numerical estimate. As a historical note, Eqn. 31 and Eqn. 36 (they differ only in parameterizations) in the referenced publication are provided without proof – it was probably obvious to Prof Metz or he wanted to leave it to us “mere mortals” to figure it out, as a final parting gesture ...

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