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2. In addition to observing living organisms, Darwin studied the preserved remains of ancient organisms called a) fossils.

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SECTION 17.1 THE FLOW OF

ENERGY—HEAT AND WORK (pages

505–510) This section explains the relationship between energy and heat, and distinguishes between heat capacity and specific heat. Energy

Transformations (page 505) 1. What area of study in chemistry is concerned with the heat transfers that

SECTION 17.1 THE FLOW OF ENERGY HEAT AND WORK (pages 505–510)

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HEAT AND WORK(pages505-510) This
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between energy and heat, and
distinguishes between heat capacity and
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Transformations(page505) 1.

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Section Review. Objectives. • Explain the relationship between energy, heat, and work. • Distinguish between exothermic and endothermic processes. • Distinguish between heat capacity and specific heat.

Vocabulary Key Equations and

Relationships. • 1 Calorie = 1 kilocalorie

1000 calories. • 1 J = 0.2390 cal and 4.184

J = 1 cal.

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- A new section on thermal energy transfer has been added to chapter 6 (Thermochemistry). The addition of this material (in section 6.4), as well as the corresponding new example and conceptual connection, gives students a better foundation for understanding calorimetry.

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