

Chapter 14 Acids And Bases

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Chapter 14 Acids And Bases

an acid that is composed of H, O, and a third element. The H & O are covalently bonded and are called the hydroxyl group. Property of bases. taste bitter, feel slippery, change color of indicators, react with acids to produce a salt and water neutralization rxn, are electrolytes. Arrhenius or traditional base.

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(14.1) The nature of acids and bases (14.2) Acid strength (14.3) The pH scale (14.4) Calculating the

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pH of strong acid solutions (14.5) Calculating the pH of weak acid solutions (14.6) Bases (14.7)
Polyprotic acids

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Chapter 14 - Acids and Bases . 14.1 The Nature of Acids and Bases . A. Arrhenius Model 1. Acids produce hydrogen ions in aqueous solutions 2. Bases produce hydroxide ions in aqueous solutions B. Bronsted-Lowry Model 1. Acids are proton donors 2. Bases are proton acceptors 3. H. 3. O + is called the hydronium ion C. Conjugate Acid- Base Pairs 1.

Chapter 14 - Acids and Bases - ScienceGeek.net

Chapter 14 - Acids and Bases. Description. Chapter 14 Review. Total Cards. 36. Subject. Chemistry. Level. 11th Grade. ... General Properties of Acids: Definition. 1) sour taste, corrosive 2) characteristically changes acid-base indicator 3) react with active metals to form hydrogen gas 4) react with bases to form salt and water 5) conduct ...

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ANS: e) ClO_4^- , F^- , OAc^- , CN^- PAGE: 14.2 11. The conjugate base of a weak acid is a) a strong base b) a weak base c) a strong acid d) a weak acid e) none of these ANS: b) a weak base PAGE: 14.2 12.

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Which of the following is the equilibrium constant expression for the dissociation of the weak acid HOCl? a) $K = \frac{[H^+][OCl^-]}{[HOCl]}$

CHAPTER 14 Acids and Bases

Modern Chemistry Chapter 14 Vocab Acids and Bases. binary acid. oxyacid. Arrhenius acid. Arrhenius base. an acid that contains only two different elements: hydrogen an.... an acid that is a compound of hydrogen, oxygen, and a third el.... a chemical compound that increases the concentration of hydrog....

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CHAPTER 14 REVIEW Acids and Bases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Name the following compounds as acids: sulfuric acid a. H_2SO_4 sulfurous acid b. H_2SO_3 hydrosulfuric acid c. H_2S perchloric acid d. $HClO_4$ hydrocyanic acid e. hydrogen cyanide 2. H

14 Acids and Bases - Password

acid, and water acts as the base. This is the acid ionization reaction. $HA(aq) + H_2O(l) \rightleftharpoons A^-(aq) + H_3O^+(aq)$ 6) When a base is added to water, there is a Bronsted-Lowry acid base reaction where the base behave as a base and water acts as the acid. This is the base ionization reaction. Be able to identify or provide the conjugate acids and ...

Chapter 14: Acids and Bases

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CHAPTER 14 ACIDS AND BASES 501 24. A Lewis acid must have an empty orbital to accept an electron pair, and a Lewis base must have an unshared pair of electrons. 25. $\text{NH}_3 + \text{NH}_3 \rightleftharpoons \text{NH}_2^- + \text{NH}_4^+$ Acid Base Conjugate Conjugate Base Acid

CHAPTER 14 ACIDS AND BASES - HCC Learning Web

Major topics: Arrhenius vs. Bronsted-Lowry definition of acids and bases, conjugate acid/base, acid dissociation constant (K_a), & strong vs weak acids

Chapter 14 (Acids and Bases) - Part 1 - YouTube

Chapter Fourteen: Acids and Bases Chapter interactives: Acids/bases and indicators Chemistry battleship Chemistry jokes pH simulation Chapter Homework: Section 1: Chapter review 1 thru 7. Section 2: Chapter review 12, 13. Section 3: Chapter review 19 thru 25.

Chapter Fourteen [Acids and Bases]

Chapter 14: Acids and Bases 1. Acids and Bases. 1. Which of the following is the equilibrium constant expression for the dissociation of the weak acid HOCl ? e) none of these. 2. In which of the following reactions does the H_2PO_4^- ion act as an acid? d) The ion cannot act as an acid.

Chapter 14: Acids and Bases 1 - Docest

Chapter 14 - Acids and Bases Pablo Gonzalez. Loading... Unsubscribe from Pablo Gonzalez? ... Chapter 16 Acid-Base Equilibria - Duration: 1:06:16. Michael Farabaugh 22,088 views.

Chapter 14 - Acids and Bases

Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In this chapter you will learn The definitions and identity of substances that are called acids Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In...

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Chapter-14-Acids-and-Bases - CHAPTER 14 ACIDS AND BASES In ...

Acids and Bases: Chapter 14 & 15 . HW: •Read Ch 14: •Fill in as much of the acid base table as you can, as you read . Acid base conductivity and reactivity Conductivity, Reactivity,, Hydrochloric*acid* high high Acidic*acid* ** low* medium Disolved*water* none* none* Ammoniumhydroxide med* none* Sodiumhydroxide* high none* *

Acids and Bases: Chapter 14 & 15

Friday, January 19 - Day A Agenda: 1) Test - Chapter 12 and 13 HW: 1) Watch Videos and take notes on intro to acids and bases , and pH scale 2) Read 14.1- 14.4

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